





# The Chemical Formulary

*Collection of Commercial Formulas  
for Making Thousands of Products  
in Many Fields*

**VOLUME XXVII**

*Editor-in-Chief*

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## PREFACE

Chemistry, as taught in our schools and colleges, concerns chiefly synthesis, analysis, and engineering—and properly so. It is part of the right foundation for the education of the chemist.

Many a chemist entering an industry soon finds that most of the products manufactured by his concern are not synthetic or definite chemical compounds, but are mixtures, blends, or highly complex compounds of which he knows little or nothing. The literature in this field, if any, may be meager, scattered, or obsolete.

Even chemists with years of experience in one or more industries spend considerable time and effort in acquainting themselves with any new field which they may enter. Consulting chemists similarly have to solve problems brought to them from industries foreign to them. There was a definite need for an up-to-date compilation of formulas for chemical compounding and treatment. Since the fields to be covered are many and varied, an editorial board of chemists and engineers engaged in many industries was formed.

Many publications, laboratories, manufacturing firms, and individuals have been consulted to obtain the latest and best information. It is felt that the formulas given in this volume will save chemists and allied workers much time and effort.

Manufacturers and sellers of chemicals will find, in these formulas, new uses for their products. Nonchemical executives, professional men, and interested laymen will make through this volume a “speaking acquaintance” with products which they may be using, trying, or selling.

It often happens that two individuals using the same ingredients in the same formula get different results. This may be due to slight deviations in the raw materials or unfamiliarity with the intricacies of a new technique. Accordingly, repeated experiments may be necessary to get the best results. Although many of the formulas given are being used commercially, many have been taken from the literature and may be subject to various errors and omissions. This should be taken into consideration. Wherever

possible, it is advisable to consult with other chemists or technical workers regarding commercial production. This will save time and money and help avoid trouble.

A formula will seldom give exactly the results which one requires. Formulas are useful as starting points from which to work out one's ideas. Also, formulas very often give us ideas which may help us in our specific problems. In a compilation of this kind, errors of omission, commission, and printing may occur. I shall be glad to receive any constructive criticism.

**H. BENNETT**

## PREFACE TO VOLUME XXVII

This new volume of the CHEMICAL FORMULARY series is a collection of new, up-to-date formulas. The only repetitious material is the introduction (Chapter I) which is used in every volume for the benefit of those who may have bought only one volume and who have no educational background or experience in chemical compounding. The simple basic formulas and compounding methods given in the introduction will serve as a guide for beginners and students. It is suggested that they read the introduction carefully and even make a few preparations described there before compounding the more intricate formulas included in the later chapters.

The list of chemicals and their suppliers has been enlarged with new trademark chemicals, so that buying the required ingredients will present no problem.

Grateful acknowledgement is made to the Contributors for their valuable suggestions and contributions.

H. BENNETT

NOTE: All the formulas in Volumes I through XXVII (except in the Introduction) are different. Thus, if you do not find what you want in this volume, you may find it in one of the others.

NOTE: This book is the result of cooperation of many chemists and engineers who have given freely of their time and knowledge. It is their business to act as consultants and to give advice on technical matters for a fee. As publishers, we do not maintain a laboratory or consulting service to compete with them. Therefore, please do not ask us for advice or opinions, but consult a chemist.



**BOOKS BY H. BENNETT**

The Chemical Formulary Vols. I—XXVI  
The Cumulative Index Vols. I—XXV—The Chemical Formulary  
Concise Chemical & Technical Dictionary  
New Cosmetic Formulary  
Chemical Specialties  
Industrial Waxes, Vols. I, II  
Practical Emulsions, Vols. I, II  
More For Your Money  
Encyclopedia of Chemical Trademarks and Synonyms, Vols. I—III

## ABBREVIATIONS

amp.....	ampere
amp/dm <sup>2</sup> .....	amperes per square decimeter
amp/ft <sup>2</sup> .....	amperes per square foot
anhyd.....	anhydrous
approx.....	approximately
ASTM.....	American Society for Testing and Materials
avoir.....	avoirdupois
bbl.....	barrel
Bé.....	Baumé
B.P.....	boiling point
°C.....	degrees Centigrade
cc.....	cubic centimeter
cd.....	current density
cm.....	centimeter
cm <sup>3</sup> .....	cubic centimeter
conc.....	concentrated
c.p.....	chemically pure
cp, cP.....	centipoise
cs.....	centistoke
cu ft.....	cubic foot
cu in.....	cubic inch
cwt.....	hundredweight
d.....	density
dil.....	dilute
dm.....	decimeter
dm <sup>2</sup> .....	square decimeter
dr.....	dram
E.....	Engler
°F.....	degrees Fahrenheit
ffc.....	free from chlorine
ffpa.....	free from prussic acid
fl dr.....	fluid dram
fl oz.....	fluid ounce
fr pt.....	flash point
F.P.....	freezing point
ft.....	foot
ft <sup>2</sup> .....	square foot
ft <sup>3</sup> .....	cubic foot
g.....	gram
gal.....	gallon
gr.....	grain
h.....	hour
hl.....	hectoliter

## ABBREVIATIONS

in.....	inch
in. <sup>3</sup> .....	cubic inch
kg.....	kilogram
l.....	liter
lb.....	pound
liq.....	liquid
m.....	meter
min.....	minim, minute
ml.....	milliliter (cubic centimeter)
mm.....	millimeter
M.P.....	melting point
MPa.....	mega pascal
N.....	Normal
N.F.....	National Formulary
oz.....	ounce
pH.....	hydrogen-ion concentration
phr.....	parts per hundred rubber
ppm.....	parts per million
psi.....	pounds per square inch
psig.....	pounds per square inch gauge
pt.....	pint
pwt.....	pennyweight
q.s.....	a quantity sufficient to make
qt.....	quart
rpm.....	revolutions per minute
s.....	second
sp.....	spirits
Sp. Gr.....	specific gravity
sq. dm.....	square decimeter
tbsp.....	tablespoon
tech.....	technical
tinc.....	tincture
™.....	trademark
T.P.....	triple pressed
tr.....	tincture
tsp.....	teaspoon
Tw.....	Twaddell
USP.....	United States Pharmacopeia
v.....	volt
visc.....	viscosity
vol.....	volume
wt.....	weight



# Chapter I

## INTRODUCTION

The following introductory matter has been included at the suggestion of teachers of chemistry and home economics.

This section will enable anyone, with or without technical education or experience, to start making simple products without any complicated or expensive machinery. For commercial production, however, suitable equipment is necessary

Chemical specialties are composed of pigments, gums, resins, solvents, oils, greases, fats, waxes, emulsifying agents, dyestuffs, perfumes, water, and chemicals of great diversity. To compound certain of these with some of the others requires definite and well studied procedures, any departure from which will inevitably result in failure. The steps for successful compounding are given with the formulas. Follow them rigorously. If the directions require that (*a*) is added to (*b*), carry this out literally, and do not reverse the order. The preparation of an emulsion is often quite as tricky as the making of mayonnaise. In making mayonnaise, you add the oil to the egg, slowly, with constant and even stirring. If you do it correctly, you get mayonnaise. If you depart from any of these details: if you add the egg to the oil, or pour the oil in too quickly, or fail to stir regularly, the result is a complete disappointment. The same disappointment may be expected if the prescribed procedure of any other formulation is violated.

The point next in importance is the scrupulous use of the proper ingredients. Substitutions are sure to result in inferior quality, if not in complete failure. Use what the formula calls for. If a cheaper product is desired, do not prepare it by substituting a cheaper ingredient for the one prescribed: use a different formula. Not infrequently, a formula will call for an ingredient which is difficult to obtain. In such cases, either reject the formula or substitute a similar substance only after a preliminary experiment demonstrates its usability. There is a limit to which this rule may reasonably be extended. In some cases, substitution of an equivalent

ingredient may be made legitimately. For example, when the formula calls for white wax (beeswax), yellow wax can be used, if the color of the finished product is a matter of secondary importance. Yellow beeswax can often replace white beeswax making due allowance for color, but paraffin wax will not replace beeswax, even though its light color seems to place it above yellow beeswax.

This leads to the third point: the use of good-quality ingredients, and ingredients of the correct quality. Ordinary lanolin is not the same thing as anhydrous lanolin. The replacement of one with the other, weight for weight, will give discouragingly different results. Use exactly what the formula calls for: if you are not acquainted with the substance and you are in doubt as to just what is meant, discard the formula and use one you understand. Buy your chemicals from reliable sources. Many ingredients are obtainable in a number of different grades: if the formula does not designate the grade, it is understood that the best grade is to be used. Remember that a formula and the directions can tell you only part of the story. Some skill is often required to attain success. Practice with a small batch in such cases until you are sure of your technique. Many examples can be cited. If the formula calls for steeping quince seed for 30 min in cold water, steeping for 1 h may yield a mucilage of too thin a consistency. The originator of the formula may have used a fresher grade of seed, or his conception of what "cold" water means may be different from yours. You should have a feeling for the right degree of mucilaginousness, and if steeping the seed for 30 min fails to produce it, steep them longer until you get the right kind of mucilage. If you do not know what the right kind is, you will have to experiment until you find out. This is the reason for the recommendation to make small experimental batches until successful results are obtained. Another case is the use of dyestuffs for coloring lotions and the like. Dyes vary in strength; they are all very powerful in tinting value; it is not always easy to state in quantitative terms how much to use. You must establish the quantity by carefully adding minute quantities until you have the desired tint. Gum tragacanth is one of those products which can give much trouble. It varies widely in solubility and bodying power; the quantity listed in the formula may be entirely unsuitable for your grade of tragacanth. Therefore, correction is necessary, which can be made only after experiments with the available gum.

In short, if you are completely inexperienced, you can profit greatly by experimenting. Such products as mouthwashes, hair tonics, and astringent lotions need little or no experience, because they are, as a rule, merely

mixtures of simple liquid and solid ingredients, which dissolve without difficulty and the end product is a clear solution that is ready for use when mixed. However, face creams, toothpastes, lubricating greases, wax polishes, etc., whose formulation requires relatively elaborate procedures and which must have a definite final viscosity, need some skill and not infrequently some experience.

### FIGURING

Some prefer proportions expressed by weight or volume, others use percentages. In different industries and foreign countries different systems of weights and measures are used. For this reason, no one set of units could be satisfactory for everyone. Thus diverse formulas appear with different units, in accordance with their sources of origin. In some cases, parts are given instead of percentage or weight or volume. On the pages preceding the index, conversion tables of weights and measures are listed. These are used for changing from one system to another. The following examples illustrate typical units.

#### Example No. 1

##### Ink for Marking Glass

Glycerin	40	Ammonium Sulfate	10
Barium Sulfate	15	Oxalic Acid	8
Ammonium Bifluoride	15	Water	12

Here no units are mentioned. In this case, it is standard practice to use parts by weight throughout. Thus here we may use ounces, grams, pounds, or kilograms as desired. But if ounces are used for one item, the ounce must be the unit for all the other items in the formula.

#### Example No. 2

##### Flexible Glue

Powdered Glue	30.90%	Glycerin	5.15%
Sorbitol (85%)	15.45%	Water	48.50%

Where no units of weight or volume, but percentages are given, forget the percentages and use the same method as given in Example No. 1.

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